The first taxane peroxide from the rooted cuttings of *Taxus canadensis*

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Abstract

A new taxane peroxide, 4α , 10β , 13α -acetoxy- 2α -benzoyloxy-7 β , 9α -epidioxy- 5β , 20-epoxytax-11-en- 1β -ol (1), was isolated and characterized for the first time in rooted cuttings of the Canadian yew, *Taxus canadensis*.

Keywords: peroxide; rooted cutting; Taxaceae; Taxane; *Taxus canadensis*; yew.

Introduction

Plants from the genus *Taxus* are a rich source of biologically active diterpenoids belonging to the unique structure class of taxanes (Baloglu and Kingston, 1999; Parmer et al., 1999). Taxol[®] (paclitaxel) is one of the most important anti-cancer drugs currently on the market against ovarian and breast cancers and shows promising effects for a variety of other cancers, such as neck, lung, gastrointestinal and bladder. Its unique mechanism of action, limited solubility and shortage have generated worldwide interest, and extensive chemical studies have

been carried out (Kingston et al., 1993; Farina, 1995; Georg et al., 1995; Baloglu and Kingston, 1999; Parmer et al., 1999; Shi and Kiyota, 2005). Taxus canadensis is a low trailing shrub ubiquitous to the Quebec region and its composition has been shown to be very different from other species (Dong et al., 2008). Interest in the Canadian yew has been prompted by the discovery that its needles can be a rich source of 9-dihydro-13acetylbaccatin III (2), 5-7 times the amount of Taxol®, depending on the collection site (Zamir et al., 1992,1995), which is specific to this variety and has only been found as traces in the bark of Taxus chinensis (Zhang et al., 1992), and has been used as a starting material to semisynthesis of Taxol[®] and its analogues (Nikolakakis et al., 2000). Previous studies on the Canadian yew have led to the identification of more than 70 taxanes including various skeletons (Dong et al., 2008). Further investigation on rooted cutting of this plant resulted in the isolation of a taxane peroxide 1 (a 7β , 9α -epidioxy derivative of 2; Figure 1). This paper reports the isolation and identification of 1. In order to compare the nuclear magnetic resonance (NMR) data especially to check the C-7-OH and C-9-OH functions, we also report the ¹H and ¹³C data for **2** in acetone- d_6 .

Results and discussion

Taxane 1 shows the molecular weight at m/z 669.2358 [M+K]⁺ and 653.2595 [M+Na]+ in high resolution-fast atom bombardment mass spectrum (HR-FABMS) analysis. ¹H and ¹³C NMR data (Table 1 and Figure 2) and heteronuclear multiple bond correaltion spectroscopy (HMBC) spectra reveal that 1 is an analog of baccatin III including three acetyls, one benzoyl, one tetrasubstituted olefin, six oxymethines, two oxygenated quaternary carbons, and four tertiary methyl groups. Two of the methyl groups at δ_{H} 1.24 and 1.26 are COSY (¹H-¹H correlation spectroscopy)-correlated peaks as geminal methyls (Me-16 and Me-17). The signals at $\delta_{\!H}$ 4.14 and 4.16 (each 1H, d, J=8.3 Hz) and $\delta_{\rm C}$ 75.7 (t) indicate the presence of an oxetane ring in the molecule. Detailed analysis of the 1H-1H COSY spectrum revealed the connectivities of C-2 to C-3, C-5 to C-7, C-9 to C-10, C-13 to C-14, and Me-18 to H-13. The cross-peaks of Me-16 and Me-17 to C-1, C-11, and C-15 in the HMBC experiment implied that Me-16 and Me-17 are attached to C-15; while the cross-peaks of Me-18 to C-11, C-12 and C-13 proved that Me-18 is connected to C-12. These HMBC correlations along with the ¹H-¹³C long-range correlations of H-14 and H-10 to C-15 suggest the presence of a cyclohexane moiety (ring A). The cross-peaks of H-2 to C-1, H-3 to C-8, H-10 to C-15 in the HMBC spectrum are indicative of the presence of an eight-membered ring (ring B). The ¹H-¹³C long-range correlations of H-3 to C-7, C-8, H-5 to C-3,



Figure 1 Structures of 1 and 2.

C-4, C-7, and H-6 to C-8 are consistent with the presence of a cyclohexane moiety (ring C). The HMBC correlations of H-20a and H-20b to C-4 and C-5 indicate that the oxetane ring is fused to the ring C at C-4 and C-5. The H-2 proton shows the long-range correlation of the carbonyl carbon at δ_{C} 165.5 in the HMBC spectrum, supporting the presence of the benzoyloxy group at C-2. The H-10 proton exhibits the three-bond correlation with a carbonyl carbon at δ_{C} 169.5 demonstrating that an acetoxy group is attached at C-10. The chemical shift of H-13 suggests that one acetoxy group is located at C-13. The remaining acetoxy group is positioned at C-4 as implied by the chemical shift of C-4 at δ_{C} 81.6, like many other taxanes containing an oxetane ring (Kingston et al., 1993; Farina, 1995). Since only one hydroxy group is detected in the ¹H NMR spectra recorded in acetone- d_6 and CDCl₃, the remaining C-7-OH and C-9-OH should be accommodated by forming a C-7-C-9 peroxide. Thus, the structure of 1 was proposed to be a 7β , 9α -

Table 1 ¹H and ¹³C NMR data for **1** in acetone- d_6 (500 MHz for ¹H, 125 MHz for ¹³C).

Position	$\delta_{H} \ (mult)^{a}$	J (Hz)	$\delta_{C}^{\ b}$	HMBC	NOESY
1	_		77.4		
1-OH	3.99 (s)			1, 2, 14, 15	2, ^w 14, ^w 16, ^w 17 ^w
2	5.86 (d)	6.2	73.5	1, 3, 8, 14, 165.5	1-OH, ^w 3, ^w 9, ^s 17, ^s 19 ^s
3	3.20 (d)	6.2	45.9	1, 2, 7, 8, 19, 20	2, ^w 7, ^s 10, ^w 14, ^m 18 ^m
4	-		81.6		
5	4.93 (d)	8.1	83.6	3, 4, 7	6α , ^s $20\alpha^{m}$
6α	2.53 (ddd)	14.9, 9.2, 7.6	36.5	7, 8	5, ^s 7 ^s
6β	1.70 (ddd)	14.9, 10.3, 1.6			7 ^m
7	4.39 (dd)	10.3, 7.6	72.1		3, ^s 6α, ^s 6β, ^m 10, ^s 18 ^s
8	-		42.1		· · · · ·
9	4.62 (d)	10.6	76.6	7, 8, 10, 19	$2,^{s} 19^{s}$
10	6.17 (d)	10.6	72.6	9, 11, 12, 15, 169.5	3, ^w 7, ^s 18 ^s
11	-		134.7		
12	_		139.5		
13	6.13 (br.t)	6.9	69.3		14, ^s 16 ^s
14αβ	2.3-2.4 (o.m)		35.9	12, 15, 1, 2, 13	1-OH, ^w 3, ^m 13 ^s
15	-		43.2		
16	1.24 (s)		27.4	1, 11, 15, 17	13, ^s 17, ^s 1-OH ^w
17	1.66 (s)		22.5	1, 11, 15, 16	1-OH, ^w 2, ^s 16 ^s
18	1.90 (d)	1.1	13.8	11, 12, 13	3, ^m 7, ^s 10 ^s
19	1.65 (s)		12.0	3, 7, 8, 9	2, ^s 9, ^s 20β ^s
20α	4.14 (d)	8.3	75.7	3, 5	5 ^m
20β	4.16 (d)	8.3		4	19 ^s
Ac	2.30 (s)		22.0	169.7	
			169.7		
	2.17 (s)		20.2	170.3	
			170.3		
	2.05 (s)		20.1	169.5	
			169.5		
BzC			165.5		
0-	8.10 (d)	8.2	129.7	165.5	
<i>m</i> -	7.52 (t)	7.5	128.3		
<i>p</i> -	7.63 (t)	7.5	133.0		

^aMultiplicity: s, singlet; d, doublet; ddd, doublet of doublets of doublets; br, broad; t, triplet; m, mutiplet; o, overlapped. ^bThe ¹³C chemical shifts were extracted from the heteronuclear multiple quantum coherence (HMQC) experiment (± 0.2 ppm). The numbers in bold represent quaternary carbons whose chemical shifts were obtained from the HMBC experiment (± 0.2 ppm). ^cNOESY intensities are marked as strong (s), medium (m), or weak (w).



Figure 2 The ¹H NMR spectrum of 1.

Table 2 ¹H and ¹³C data for 9-dihydro- 13α -acetylbaccatin III (**2**) in acetone- d_6 (500 MHz for ¹H, 125 MHz for ¹³C).

Position	δ (H) mult	J (Hz)	$\delta\left(C\right)$	HMBC	NOESY
1	_		77.4		
1-OH	3.88 (s)			1, 2, 14, 15	2, ^w 14, ^w 16, ^w 17 ^w
2	5.79 (d)	6.0	73.4	1, 3, 8, 14, 15, 165.5	1-OH, ^w 3, ^w 9, ^s 17, ^s 19 ^s
3	3.07 (d)	6.0	46.9	1, 2/7, 8, 19, 20	2, ^w 7, ^s 14, ^s 18 ^m
4	_		81.6		
5	4.91 (d)	8.0	83.6	3, 4, 7	6α, ^s 20α ^w
6α	2.45 (ddd)	14.9, 9.1, 7.7	37.7	5, 7, 8	5, ^s 7 ^s
6β	1.81 (o.m)				
7	4.42 (br.m)		73.4		3, ^s 6α, ^s 10, ^s 18, ^s 7-OH ^s
7-OH	5.58 (d)	2.8			7, ^s 19 ^w
8	-		44.8		
9	4.48 (br.dd)	10.4, 7.7	76.6	7/10, 8, 19	2, ^s 17, ^s 19, ^m 9-OH ^m
9-OH	5.74 (d)	7.3		8,9	9, ^m 10 ^w
10	6.19 (d)	10.8	72.8	9, 11, 12, 15, 169.5	3, ^w 7, ^s 9-OH, ^m 18 ^s
11	-		135.8		
12	-		138.1		
13	6.13 (br.t)		69.3		14, ^s 16 ^s
14αβ	2.3-2.4 (o.m)		35.9		1-OH, ^w 3, ^s 13 ^s
15	-		43.2		
16	1.22 (s)		27.4	1, 11, 15, 17	13, ^s 17, ^s 1-OH ^w
17	1.66 (o.s)		22.5	1, 11, 15, 16	1-OH, ^w 2, ^s 9, ^s 16 ^s
18	1.92 (d)	1.3	13.8	11, 12, 13	3, ^m 7, ^s 10 ^s
19	1.78 (s)		12.0	3, 7, 8, 9	2, ^s 7-OH, ^w 9, ^s 9-OH, ^w 20β ^s
20α	4.11 (d)	8.2	75.7	3, 5	5 ^w
20β	4.17 (d)	8.2		4	19 ^s
OAc	2.32 (s)		22.0	169.7	
			169.7		
	2.17 (s)		20.2	170.3	
			170.3		
	2.06 (s)		20.1	169.5	
			169.5		
BzC				165.5	
0-	8.10 (d)	8.3	129.7		
<i>m</i> -	7.52 (t)	7.6	128.3		
<i>p</i> -	7.63 (t)	7.5	133.0		



Figure 3 The ¹H NMR spectra of the mixture of 1 and 9-dihydro- 13α -acetylbaccatin III (2).

epidioxy derivative of 9-dihydro-13 α -acetylbaccatin III (2), though the HR-FABMS data of 1 showed the same molecular formula with 2. Indeed, the ¹H NMR data of 1 closely resemble those of 2 (Table 2 and Figure 3) (Zamir et al., 1995) except that H-2 and H-3 show slight upfield shifts for 1. As the original NMR data of 1 and 2 were recorded in CDCl₃ solution, in order to address whether C-7-OH and C-9-OH were involved in a peroxide ring, we re-run NMR data in acetone- d_6 , and not C-7-OH and C-9-OH signals but the C-1-OH signal was observed for 1. On the other hand, signals for all three free hydroxy groups were observed for 2. This fact further substantiates our presumed structure. The relative stereochemistry of 1 was elucidated by the nuclear Overhauser effect correlated spectroscopy (NOESY) spectrum as shown in Figure 4, which



Figure 4 Relative stereochemistry of 1. Dotted arrows denoted NOESY correlations.

is identical to that of **2**. This taxane, 4α , 10β , 13α -acetoxy- 2α benzoyloxy- 7β , 9α -epidioxy- 5β , 20-epoxytax-11-en- 1β -ol (**1**), is the first example of a taxane peroxide.

Conclusion

 4α , 10β , 13α -acetoxy- 2α -benzoyloxy- 7β , 9α -epidioxy- 5β , 20-epoxytax-11-en- 1β -ol (1), the first example of a taxane peroxide, was isolated from rooted cuttings of the Canadian yew, *Taxus canadensis*.

Experimental

General

Optical rotation: Jasco DIP-370; NMR: Bruker Avance-500; FAB-MS: Vacuum Generators ZAB-HS; flash chromatography: Silica Gel 60 (230–400 mesh, EM Science); preparative thin-layer chromatography (TLC): Silica Gel 60 F_{254} (20×20 cm×0.5 mm, EM Science); preparative HPLC: Waters Delta Prep 3000, UV 486 (210 nm), Whatman partisil 10 ODS-2 Mag-9 (9.4×250 mm).

Plant material

Rooted cuttings of *T. canadensis* Marsh. (Taxaceae) were obtained in September 2001 from Cramer Nurseries in Les Cèdres, Québec, Canada.

Air-dried rooting cuttings of T. canadensis were ground into a fine powder (584 g) which was then extracted with 3 l of MeOH by shaking at room temperature for 1 d in the dark. The solvent was removed and the plant material was extracted twice with MeOH and four times with a mixture of MeOH/ CH₂Cl₂ (1:1). The combined organic extracts were concentrated to dryness and the residue was re-dissolved in water. Then the aqueous solution was extracted with hexane. The hexane fraction was partitioned three times with MeOH. The combined MeOH extracts were dried over anhydrous sodium sulphate, filtered and evaporated to dryness, yielding 2.0 g of a dark brown extract. The extract was re-dissolved in a minimum volume of CH2Cl2 and fractionated by column chromatography (180 g, bed size 4.0×33 cm). Twenty-three fractions (Fr₁₋₂₃) were collected after elution with successive mixtures of acetone in hexane. Fr₁₋₅ to Fr₁₋₇ were combined (60 mg) and applied to preparative TLC. Development with hexane/ethyl acetate (2:5) yielded crude taxane 1 (3.0 mg, $R_{=}$ =0.30), which was further purified by preparative HPLC, eluted with a linear gradient of acetonitrile in water from 25% to 100% in 50 min at 3 ml/min. The material eluted at $t_p=30.7$ min was collected as a pure taxane 1; amorphous solid; $[\alpha]_D^{22}$ +27.5 (c 0.04, CHCl₃); for ¹H- and ¹³C NMR spectral data in acetone- d_6 , see Table 1; ¹H NMR in CDCl₃: δ 5.82 (1H, d, J=6.4 Hz, H-2), 3.19 (1H, d, J=6.4 Hz, H-3), 4.94 (1H, br.d, J=9.3 Hz, H-5), 2.63 (1H, ddd, J=15.2, 9.3, 7.4 Hz, H-6α), 1.85 (1H, ddd, J=15.1, 10.5, 1.5 Hz, H-6β), 4.42 (1H, dd, J=10.5, 7.4 Hz, H-7), 4.61 (1H, d, J=10.6 Hz, H-9), 6.15 (1H, d, J=10.6 Hz, H-10), 6.14 (1H, m, H-13), 2.20 (2H, d, J=8.4 Hz, H-14αβ), 1.25 (3H, s, Me-16), 1.67 (3H, s, Me-17), 1.91 (3H, d, J=1.3 Hz, Me-18), 1.67 (3H, s, Me-19), 4.31 (1H, d, J=8.4 Hz, H-20a), 4.15 (1H, d, J=8.4 Hz, 20b), 2.28 (3H, s, Ac), 2.18 (3H, s, Ac), 2.13 (3H, s, Ac), 8.08 (2H, d, J=8.3 Hz, Ph-o), 7.48 (2H, d, J=7.6 Hz, Ph-m), 7.61 (1H, t, J=7.5 Hz, Ph-p), 3.32 (1H, br.s, 1-OH); ¹³C NMR in CDCl₃: δ 78.8 (C-1), 73.5 (C-2), 45.9 (C-3), 81.9 (C-4), 84.0 (C-5), 36.1 (C-6), 72.8 (C-7), 44.6 (C-8), 77.2 (C-9), 72.7 (C-10), 134.0 (C-12), 69.8 (C-13), 35.4 (C-14), 42.9 (C-15), 27.8 (C-16), 22.4 (C-17), 15.0 (C-18), 12.7 (C-19), 76.3 (C-20), 22.8 (CH₃CO), 21.2 (CH₃CO), 21.1 (CH₃CO), 169.3 (CH₃CO), 170.2 (CH₃CO), 170.3 (CH₃CO), 129.9-128.3 (Ph); HR-FABMS m/z: 669.2358 (calcd for $C_{33}H_{42}O_{12}K$, 669.2308), 653.2595 (calcd for C33H42O12Na, 653.2569), 593.2375 (calcd for $[C_{33}H_{42}O_{12}Na-CH_3COOH]^+$, 593.2357), 533.2158 (calcd for [C₃₃H₄₂O₁₂Na-2CH₃COOH]⁺, 533.2146), 473.1950 (calcd for [C₃₃H₄₂O₁₂Na-3CH₃COOH]⁺, 473.1935), $411.1783 (calcd for [C_{33}H_{42}O_{12}Na-2CH_{3}COOH-PhCOOH]^{+},$ 411.1778), 351.1574 (calcd for [C₃₃H₄₂O₁₂Na–3CH₃COOH–PhCOOH]⁺, 351.1567).

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